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The submission has been updated. Below is the information submitted.

Abstract Title: Solubility Calculation of CO2 or H2 in Complex Polar Liquid

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Options:

- Fundamentals - Thermodynamics and transport properties (experiment and modelling)
- Fundamentals - Phase equilibria (experiment and modelling)

Abstract: When the critical properties of a pure substance are known, it is possible to predict thermodynamic properties using the c
For the calculation of mixture properties, best fitted binary interaction parameter, m_{ij} , is effective. Especially, for a system c
In this work, using the developed Joback method, the critical temperature of triolein and ionic liquids which are glymes and TCMS a
Hydrogen solubility in triolein has temperature dependent ($m_{ij}=-1.8$) in comparison with the correlated m_{ij} value -1.16 [5] as show
In conclusion, solubility of CO2 or H2 in complex polar liquids can be calculated, or sometimes predicted using an equation of sta

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Keywords: Solubility Calculation, CO2, H2, Polar Liquid

Comments:

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